

Co-Fe-Ho-Sm (Cobalt-Iron-Holmium-Samarium)

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Recently, [2004Liu] determined an isothermal section for this system at 800 °C and at a constant Ho/Sm atom ratio of 1.

Binary Systems

In the Co-Fe system, the face-centered cubic (fcc) Fe forms a continuous solid solution γ with α Co over a wide range of temperature. The γ phase transforms congruently to the body-centered cubic (bcc) solid solution α at 985 °C and 45 at.% Co. The Co-Ho phase diagram up to 33.3 at.% Ho depicts the following compounds: $\text{Co}_{17}\text{Ho}_2$ (Th₂Zn₁₇-type rhombohedral), Co_5Ho (CaCu₅-type hexagonal), Co_7Ho_2 (Co₇Er₂-type rhombohedral), Co_3Ho (Ni₃Pu-type rhombohedral) and Co_2Ho (MgCu₂-type cubic). The Co-Sm diagram up to 33.3 at.% Sm depicts the following compounds: $\text{Co}_{17}\text{Sm}_2$ (Th₂Zn₁₇-type rhombohedral), Co_5Sm (CaCu₅-type hexagonal), $\text{Co}_{19}\text{Sm}_5$, Co_7Sm_2 (Co₇Er₂-type rhombohedral), Co_3Sm (Ni₃Pu-type rhombohedral), and Co_2Sm (MgCu₂-type cubic). The Fe-Ho phase diagram depicts four intermediate compounds: $\text{Fe}_{17}\text{Ho}_2$ (Th₂Ni₁₇-type hexagonal), $\text{Fe}_{23}\text{Ho}_6$ (Mn₂₃Th₆-type cubic), Fe_3Ho (Ni₃Pu-type rhombohedral), and Fe_2Ho (MgCu₂-type cubic). The Fe-Sm phase diagram depicts the following phases: $\text{Fe}_{17}\text{Sm}_2$ (Th₂Zn₁₇-type rhombohedral and Th₂Ni₁₇-type hexagonal), Fe_3Sm (Ni₃Pu-type rhombohedral) and Fe_2Sm (MgCu₂-type cubic).

Ternary Systems

In the Co-Fe-Ho system, Co_2Ho and Fe_2Ho form a continuous MgCu₂-type cubic solid solution. In the

Co-Fe-Sm system, the pairs $\text{Co}_{17}\text{Sm}_2$ - $\text{Fe}_{17}\text{Sm}_2$, Co_3Sm - Fe_3Sm , and Co_2Sm - Fe_2Sm form continuous solid solutions.

Quaternary Isothermal Section

With starting metals of 99.99% Co, 99.8% Fe, 99.8% Ho, and 99.8% Sm, [2004Liu] prepared by arc-melting 5 g alloy samples, which were annealed at 800 °C. The phase equilibria were studied by x-ray powder diffraction. The isothermal section constructed by [2004Liu] at 800 °C and at a constant Ho/Sm atom ratio of 1 is redrawn in Fig. 1. R = (Ho + Sm) in Fig. 1. According to the (Ho,Sm):(Fe,Co) ratio, the quaternary phases are denoted as 1:2 (MgCu₂-type), 1:3 (Ni₃Pu-type), 1:5 (CaCu₅-type), 2:17(r) (Th₂Zn₁₇-type), 2:17(h) (Th₂Ni₁₇-type), 2:7 (Co₇Er₂-type), and 6:23 (Mn₂₃Th₆-type). The 1:2 and 1:3 phases form continuous solid solutions. Fe_{23}R_6 and Fe_{17}R_2 dissolve Co to the extent of $(\text{Fe}_{0.9}\text{Co}_{0.1})_{23}\text{R}_6$ and $(\text{Fe}_{0.8}\text{Co}_{0.2})_{17}\text{R}_2$, respectively. Co_7R_2 , Co_5R and Co_{17}R_2 dissolve Fe up to $(\text{Co}_{0.77}\text{Fe}_{0.23})_7\text{R}_2$, $(\text{Co}_{0.95}\text{Fe}_{0.05})_5\text{R}$, and $(\text{Co}_{0.3}\text{Fe}_{0.7})_{17}\text{R}_2$ respectively. The phase corresponding to 5:19 was not found by [2004Liu].

Reference

2004Liu: W. Liu, D. Sun, and G. Jin, A Study of the Isothermal Phase Change of (Fe-Co-(Sm_{0.5}Ho_{0.5}) Series Alloys, *J. Shenyang Inst. Technol.*, 2004, **23**(1), p 60-62, in Chinese

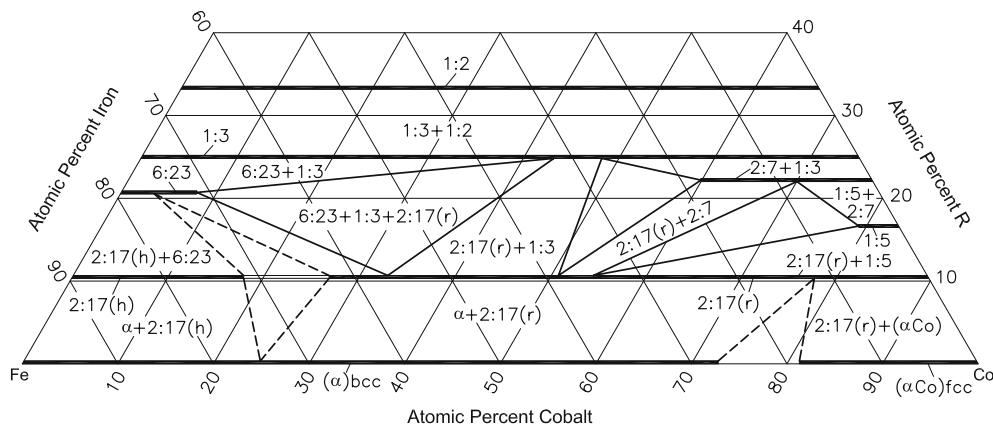


Fig. 1 Co-Fe-Ho-Sm isothermal section at 800 °C and at Ho/Sm = 1 [2004Liu]